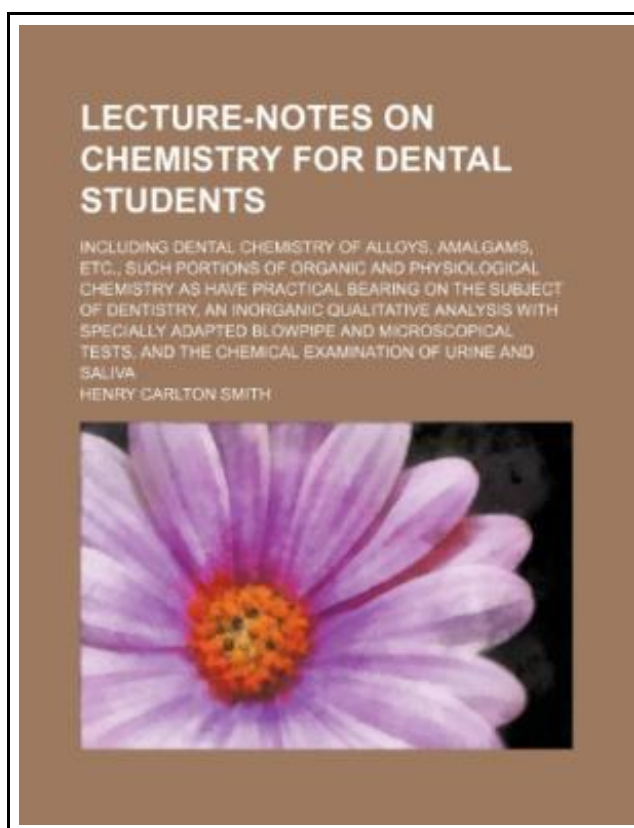


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**LECTURE-NOTES ON CHEMISTRY FOR DENTAL STUDENTS;
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SUCH PORTIONS OF ORGANIC AND PHYSIOLOGICAL CHEMISTRY AS
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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1906 edition. Excerpt: . in the addition of oxygen but in the withdrawal of hydrogen; thus the oxidation of methyl alcohol produces formaldehyd (CH₂O) and water. CH₃OH + O = CH₂O + H₂O. Aldehyds may be considered compounds containing an H H / I alkyl radical and a distinctive group, -C; thus CHO is for O maldehyd, CH₃, in acetaldehyd, etc. (compare Alcohol, page CHO 131). Formaldehyd coagulates albumen and hardens gelatine; when used as a preservative it renders the proteids tougher and less digestible. Formaldehyd polymerizes, producing the paraform or paraformaldehyd of trade, trioxymethylene, with a probable formula of (CH₂O)₃. It also forms one lower polymer (CH₂O)₂ and at least one higher, formose, a substance allied to glucose. Acetaldehyd, aldehyd, CH₃-CHO or C₂H₄O, the aldehyd from ethyl alcohol, may be made by addition of H₂SO₄ to a mixture of alcohol and bichromate of potassium. It is a colorless, inflammable liquid with pungent ethereal odor and boils at 22 C. Paraldehyd, (C₂H₄O)₃, a polymer of acetic aldehyd, is a colorless liquid with a strong pungent odor, soluble in 8.5 parts of water at 15 C, miscible in all proportions with alcohol, ether, and fixed or volatile oils. (U. S. P.) It is a valuable hypnotic. Chloral, CCl₃CHO, trichloraldehyd, is an oily liquid formed by action of dry C₁ gas on pure alcohol; soluble in ether and chloroform, boiling at 94 C to 98 C, and forming, with a molecule of H₂O chloral hydrate, CCl₃CHO.H₂O, a crystalline solid, and this is...



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